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~~Chemistry~~

~~Reactions and~~

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~~Molarity~~

~~Problems and~~

~~Examples How to~~

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~~problems (ACER~~

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~~Molarity Made~~

~~Easy: How to~~

~~Calculate~~

~~Molarity and~~

~~Make Solutions~~

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problem with
chiral~~

~~stereocentres
for Vitamin B12~~

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Phase Diagrams

of Water \u0026amp;

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rnL at a

pressure of 101 kPa and

temperature of 295 K. When the

pressure is changed, the

volume becomes

477 mL. If there has been no

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change in
temperature,
what is the new
pressure? A gas
occupies a
volume of 2.45 L
at a pressure of
1.03 atm and a
temperature of
293 K.

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Problems

the factory is

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761.3 mm Hg. The pressure inside the bag is the same. What is the pressure inside the bag of potato Chips in Pa? The same bag of potato chips from Problem 5 is shipped to a town in Colorado, where

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the atmospheric pressure is 99.82 kPa. What is the difference (in Pa) between the pressure in the bag and the atmospheric

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kPa and

temperature of

295 K. When the

pressure is

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Volume becomes
477 mL. If there
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problems is
changed, the
volume becomes
477-mL. If there
has been no
change in
temperature,
what is the new
pressure? 6.

What will be the
volume of a gas
sample at 309-K
if its volume at
215-K is 3.42-L?

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of gases
discussed in
Chapter 10 to
explain why the
solubility of
virtually all
gases in liquids
decreases with
increasing
temperature.

13.E: Properties
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Problems 1. What

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Volume would be occupied by 100. g of oxygen gas at a pressure of 1.50 atm and a temperature of 250C?

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*273k) 2. An air-filled balloon has a volume of 5 L at 0.94 atm and 250C. Soon after, the pressure changes

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to 0.99 and the
temperature
changes to 0°C.
What is the new
volume of the
balloon? / .26
1.26 L ...

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