

## Thermochemistry Review Worksheet And Answers

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Answer the following questions with regards to the Heating Curve. 1) What is happening to the average potential energy of the molecules in the sample during section 3? Potential Energy remains constant. It only increases or decreases when temperature is remains constant on the plateaus. ... Thermochemistry Review Worksheet ...

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Thermochemistry Review Worksheet. True or False If the answer is false, you must explain why. 1. \_\_\_\_ Chemical energy is a form of kinetic energy. 2. \_\_\_\_ When we mix water at 100 C with water at 20 C the final water temp. is always 60 C. 3. \_\_\_\_ When we put a 100g piece of metal at 100 C into 100g of water at 20 C the final temp is 20 C. 4.

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Thermochemistry is study of changes in energy (heat) associated. with physical or chemical changes. Force = push  $F = m a$  (mass x acceleration) force units: N (newton) = kg m s<sup>-2</sup>. Work = force x distance  $W = F d$ . energy units: J (joule) = kg m<sup>2</sup> s<sup>-2</sup>. Energy is the capacity to do work.

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A 100 gram sample of a metal undergoes a temperature change from 20 ° C to 50 ° after absorbing 1500 J of heat. What is the specific heat of the metal?

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Thermochemistry Worksheet 1 Answers Thermochemistry Answers - Worksheet Number One. We will ignore any heats losses to the walls of the container and losses to the air. These is a typical position to take since, in a real experiment, both would have to be accounted for, making for much more complexity. 1. q = (20.0 g) (20.0 ° C) (2.02 J/g ° C).

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